182. The Enantioselective Synthesis of 3-Methoxy-1,3,5(10)-estratrien-11,17-dione by a Thermal Intramolecular Cycloaddition Reaction¹)

Preliminary Communication

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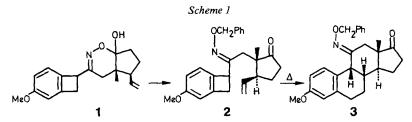
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Summary

The enantiomerically pure (+)-3-methoxy-1,3,5(10)-estratrien-11,17-dione 11 (with *trans-anti-trans* configuration) was synthesized in a highly stereocontrolled fashion from (\pm) -t-butyl 4-methoxy-1-benzocyclobutene carboxylate (8) and the (+)-carboxylic acid 6, obtained from 4 in two steps, followed by one crystallization of the (+)-ephedrine salt. The key step $10 \rightarrow 11$ (Scheme 2) involves a thermal intramolecular cycloaddition reaction.

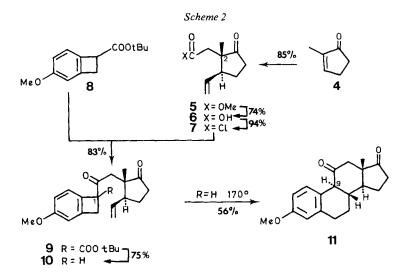
In continuation of our work on intramolecular o-quinodimethane cycloadditions²) [1] [2] a stereoselective synthesis of the racemic 3-methoxy-1,3,5(10)-9 β estratrien-11,17-dion-11-oxime-o-benzyl ether 3 (with *cis-anti-trans* configuration) via the intermediates 1 and 2 has been developed (Scheme 1) [3]. In view of the efficient and highly stereocontrolled preparation of 1 this synthesis compares favorably with other approaches to racemic estrone derivatives via benzocyclobuteneprecursors [4].



We now wish to report a related, even simpler and moreover enantioselective route to C(11) functionalized steroids as follows (*Scheme 2*). Conjugate addition of lithium 3,3-dimethylbutynyl-vinylcuprate, followed by alkylation of the nonisolated

¹) Presented in part by one of us (W.O.) at the Symposium on 'Strategy in Organic Synthesis', Wageningen, Holland, March 1978.

²) Reviews: [2].



enolate adduct with methylbromoacetate³) [5] furnished a 6:1 mixture of the ester 5^4) and its C(2)-epimer in 85% yield.

Saponification of this ester mixture $(0.02 \times \text{KOH}$ in MeOH/H₂O 10:1, 21 h/20°), followed by one crystallization of the crude (\pm) -acid with (+)-ephedrine from ethyl acetate gave a salt (m. p. 166–169°, $[a]_D^{22°} = +60.9°$ (c = 0.55, methanol)), which after treatment with hydrochloric acid afforded the enantiomerically pure (+)-acid **6**⁴) (oil, $[a]_D^{22°} = +74.6°$ (c = 0.67, CHCl₃) in 74% yield (from crude (\pm) -5). Reaction of (+)-**6** with (COCl)₂ (3 mol-equiv. in dry CH₂Cl₂/0° to 20°) furnished the (+)-acid chloride 7⁴) (b. p. 110° (bath)/0.01 Torr, $[a]_D^{22°} = +71.3°$ (c = 0.47, CHCl₃) 94% yield). The racemic benzocyclobutenecarboxylic ester **8**⁴), prepared from the corresponding acid in 73% yield⁵) [9], was treated successively with lithium cyclohexylisopropylamide (1.05 mol-equiv.) and (+)-7 (1.0 mol-equiv.) in THF at -78° [10] to give the diketoester **9**⁴) ⁶) in 83% yield.

⁴) The IR. and ¹H-NMR. spectra (100 MHz) and MS. of this compound are in agreement with the assigned structure.

⁵) A solution of 4-methoxybenzocyclobutene-1-carboxylic acid [4a,b] (15 mmol) in acetonitrile (8 ml) was added to the reaction mixture, obtained from DMF (6 ml) and oxalylchloride (22 mmol) in acetonitrile (4 ml) at -20°. After addition of pyridine (6 ml) and *t*-butyl alcohol (50 mmol) at -30° to -20° the mixture was warmed to 20°, to give after work-up and distillation 8 (b.p. 160° (bath)/ 0.01 Torr), 73%). We thank Dr. P. Stadler/Sandoz AG for kindly communicating to us this esterification method prior to publication [9].

³) 2-Methyl-2-cyclopentenone [6] (20.5 mmol) was added at - 70° to a solution of lithium 3,3-dimethylbutynyl cuprate [7] (prepared from t-butylacetylene [8], 20.4 mmol) in ether/THF 9:1 (130 ml). Then the mixture was successively allowed to warm to - 10°, cooled to - 30° and transferred by argon pressure over a 10 min period to a stirred and cooled (- 18°) solution of methyl bromoacetate (178 mmol) in HMPT/ether 2:1 (150 ml). After warming of the resulting solution to 20° over a 2 h period, stirring for 18 h and subsequent shaking with sat. aq. NH₄Cl/ether and distillation of the dried and evaporated ether solution gave 5 (b.p. 57-58°/0.04 Torr, 85%).

⁶) 1:1-Mixture of C(1)-epimers which is of little synthetic relevance since the chirality of C(1) is lost in the thermal opening of the benzocyclobutene **10** [1,2].

Removal of the COOtBu-group from 9 [a) CF₃COOH/0°, b) DMF/pyridine 30:1, 90°/30 min] furnished the 1,4-diketone 10^4)⁶) in 75% yield. An alternative route to racemic 10 involves treatment of 1 with ceric ammonium nitrate (2 molequiv.) in MeOH/H₂O 9:1 (20°, 5 h, 32% yield)⁷).

After heating a solution of 10 (0.33 mmol) in decane (10 ml) under a pure argon atmosphere at 150° for 12 h the optically pure (+)-3-methoxy-1,3,5(10)-estradien-11, 17-dione 11⁴) (with *trans-anti-trans* configuration) (m. p. 180-192°, $[a]_D^{22°} = +387°$ $(c=0.5, CHCl_3)$) crystallized directly from the cooled reaction mixture (52%). The mother liquor gave, after chromatography (SiO₂) and crystallization, another crop of pure 11 (total yield of 11: 56%) followed by a small amount of its *cis-anti-trans*-C(9)-epimer (5% yield). The synthetic (+)-11 was shown to be identical with an authentic sample⁸) by chiroptic, spectral, chromatographic and mixed m.p. evidence. It thus appears that during the reaction $10 \rightarrow 11$ the intermediate *trans*-quinodimethane [1] [2] adds preferentially in an *exo* mode to the vinyl group as opposed to the exclusive *endo* addition observed in the transformation $2\rightarrow 3^9$). Accordingly the correct induction of all three chiral centers in the key step $10 \rightarrow 11$ of this convergent synthesis, together with the selective functionalizability of the C(11)-carbonyl group in 11 to give physiologically active products [12], nicely exemplifies the utility of intramolecular *o*-quinodimethane cycloadditions.

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⁹) For related possibilities of directing intramolecular cycloadditions towards either *endo-* or *exo-* products see [2].

⁷) The feasibility of the cleavage $1 \rightarrow 10$ has been demonstrated in context with another synthetic target [11].

⁸) We are indebted to Dr. *M.G. Buzzolini*, Sandoz Inc., Hannover, N.J. for kindly providing chiral 3-methoxy-1,3,5(10)-estratrien-11,17-dione-17-ethyleneacetal, which by acetal cleavage (0.9N HClO₄ in THF/H₂O 3.5:1.5/20°/18 h) gave 11 of natural origin $[a]_{22}^{22} = +364^{\circ} c = 0.5$, CHCl₃).